

Lectures 21-22

Solid state materials. Electronic structure and conductivity

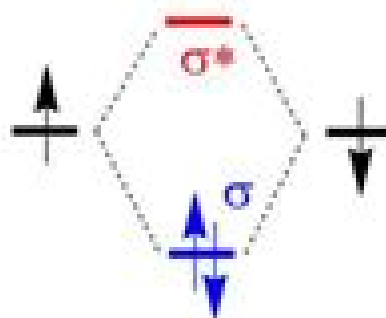
1) Band theory

- The electronic structure of **solids** can also be described by MO theory.
- A solid can be considered as a **supermolecule**.
- One mole of atoms (N_A), each with X orbitals in the valence shell contributes X moles of atomic orbitals producing X moles of MO's.

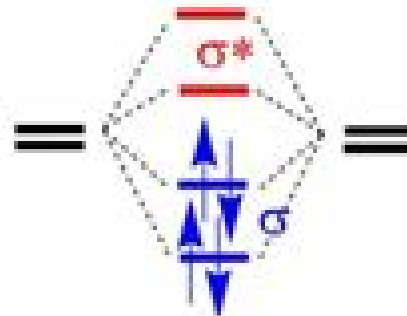
Consider qualitatively bonding between N *metal atoms* of ns^1 configuration (Li, Na etc) arranged in a *chain*; $N = 2, 4, N_A$. Assume that $X=1$ for simplicity.

- In the case of $N \sim N_A$ atoms they form not **bonds** but **bands**.
- The band appearing in the *bonding* region is called **valence band**. The *antibonding* region is called **conduction band**.
- In the case of metals the **valence** and **conduction** bands are immediately **adjacent**.

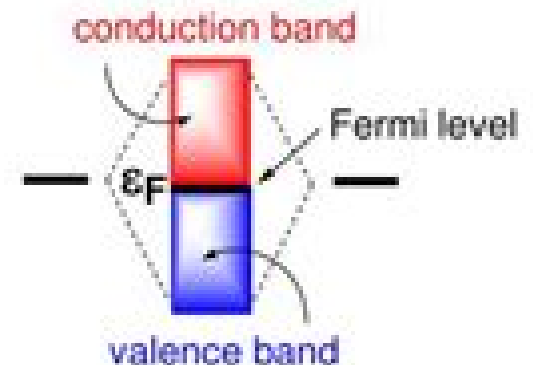
$N = 2$ Li atoms



4 Li atoms



N_A Li atoms



Atomic And Electronic Structure Of Solids

Michel Lannoo, Paul Friedel



Atomic And Electronic Structure Of Solids:

Atomic And Electronic Structure Of Solids Book Review: Unveiling the Power of Words

In a world driven by information and connectivity, the energy of words has become much more evident than ever. They have the capacity to inspire, provoke, and ignite change. Such may be the essence of the book **Atomic And Electronic Structure Of Solids**, a literary masterpiece that delves deep into the significance of words and their effect on our lives. Published by a renowned author, this captivating work takes readers on a transformative journey, unraveling the secrets and potential behind every word. In this review, we shall explore the book's key themes, examine its writing style, and analyze its overall effect on readers.

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