

1. The infinite-dilution activity coefficients in the methanol(1)/chloroform(2) binary at 25°C are

$$\gamma_1^\infty = 9.55 \text{ and } \gamma_2^\infty = 2.34.$$

(Landau, et al. *Ind. Eng. Chem. Res.* 1991;30:1900-1906.) Pure-component densities at 25°C and solubility parameters appear below.

Species	ρ @ 25°C (g/cm ³)	δ (cal/cm ³) ^{1/2}
CH ₃ OH	0.792	14.5
CHCl ₃	1.489	9.21

Prepare a table in which you report the values of activity coefficients γ_1 and γ_2 at 25°C vs x_1 for $0 \leq x_1 \leq 1$ (in increments of 0.1), computed using the following methods:

- The two-constant Margules (2CM) model, parameterized using γ_1^∞ and γ_2^∞ ;
- The van Laar (vL) model, parameterized using γ_1^∞ and γ_2^∞ ;
- Regular Solution Theory (RST), using the given data on ρ and δ ;

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Yicheng Fang



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