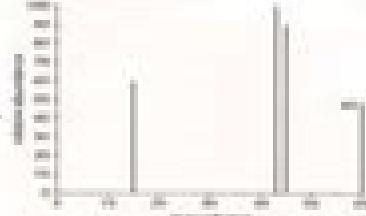
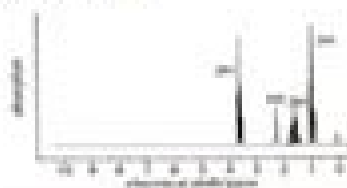
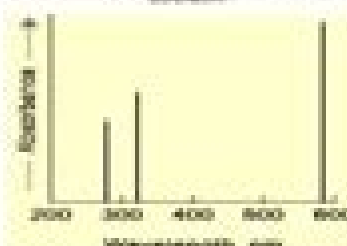
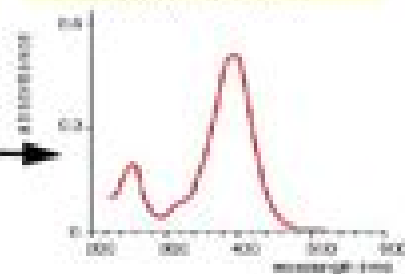


NAME	EM radiation	WHAT IS MEASURED	HOW	
Mass spectroscopy	none	Measures structure of molecules	Molecules are split up into parts and their mass is measured.	
Chromatography	none	Separation & identification of mixtures	Different constituents travel at different speeds which makes them separate.	Paper Thin Layer Column High Performance Liquid Gas Liquid
NMR spectroscopy	radiowaves $\lambda = 0.5-75 \text{ m}$	Measures structure of molecules	Radiowaves are absorbed when protons go from one spin state to another.	
IR spectroscopy	Infrared $\lambda = 2.5-15 \mu\text{m}$	Measures structure of molecules	Molecules with a polar bond are made to vibrate by IR radiation which is absorbed.	
Atomic Absorption spectroscopy	Visible + UV $\lambda = 190-900 \text{ nm}$	Measures concentrations of atoms	Atoms absorb light when electrons are excited.	
UV-vis spectroscopy	Visible + UV $\lambda = 190-900 \text{ nm}$	Measures concentrations of molecules	Molecules absorb light when d-electrons are excited in metal complexes and π -electrons are excited in organic molecules.	

Analytical Techniques In Occupational Health Chemistry

K Morrison



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