

The circular economy model:
less raw material, less waste, fewer emissions



Tutorial Circular Economy

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web suggested activity use this resource could be used as a whole class investigation with older children observing differences and similarities with hand warmers leading to research on how they work children can investigate why materials have been chosen for the hand warmers linking to previous knowledge on materials and properties

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web $m \times c \times \Delta t$ equation 1 where m is the total mass of the solution solute plus solvent c is the specific heat of the solution and Δt is the observed temperature change the specific heat of the solution is generally assumed to be the same as that of guided inquiry designing the ideal hand warmer - Dec 27 2022

web the heat of reaction ΔH_{soln} is written after the products in units of kJ mol rxn for example sodium hydroxide dissolves exothermically releasing 44.2 kilojoules per mole dissolved this process is represented as $\text{NaOH (s)} \rightarrow \text{NaOH (aq)}$ $\Delta H_{\text{soln}} = -44.2 \text{ kJ mol rxn}$ write equations to similarly represent the dissolving process for

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web in this lab you are challenged to use chemistry to determine which ionic compound makes the best hand warmer in this experiment you will collect data that will allow you to calculate the heat of solution ΔH_{soln} or the amount of energy released or absorbed in kJ/mol of the substance being dissolved

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