# Chapter 1: Chemical Basis of Life

#### Four Major Biomolecules

- 1. Amino Acids: contain an amino group and a carboxylic group and an side chain/R group
  - Under physiological conditions, ionized to NH<sub>3</sub>\* and COO
  - Ex. Alanine and Asparagine
- II. Carbohydrates: (monosaccharides/sugars)
  - Formula = (CH<sub>2</sub>O)<sub>=</sub>
  - Ex. Glucose
- III. Nucleotides: five-carbon sugar, nitrogen-containing ring, and one or more phosphate groups.
  - Ex. Adenosine Triphosphate (ATP)
- Lipids: poorly soluble in water due to its bulky hydrocarbon-like structure.
  - Ex. Palmitate and Cholesterol

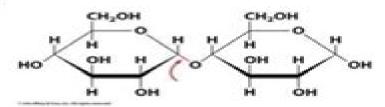
#### 3 Major Biological Polymers

- Proteins (Polypeptides): Polymers of amino acids.
  - Amino acids linked via peptide bonds.
  - Can perform tasks such as mediating chemical and provide structural support.

$$H_3\stackrel{+}{N} - \stackrel{R_1}{C} - \stackrel{0}{C} - \stackrel{R_2}{N} - \stackrel{0}{C} - \stackrel{reactions}{C}$$

- Nucleic Acids (Polynucleotides): polymers of nucleotides;
   RNA
  - Primary Role: carriers of genetic information, contained in sequence of nucleotide residues rather than their 3D
  - RNA: nucleotides adenine, cytosine, guanine, and uracil
  - DNA: nucleotides adenine, cytosine, guanine, and (ACGT)

- DNA and their shape. (ACGU) thymine
- III. Polysaccharides; contain only one or a few different type of monosaccharide residue
  - Limited potential for carrying genetic information in the sequence of their residues or for adopting a large variety of shapes and mediating chemical reactions
  - Serve as fuel-storage molecules and provide structural support.
  - Linked via glycosidic bonds



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**Clemens Wendtner** 

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