IMPORTANCE OF PHYSIOCHEMICAL PHENOMENA IN BIOCHEMISTRY:

The general laws and principles of chemistry and physics are applicable to biochemistry as well. It is, therefore, worthwhile to have a brief understanding of some of the basic chemical and physical principles that have direct relevance to life.

Most properties and laws of matter are applicable to biological systems. Some of such important properties are discussed below.

COLLOIDAL STATE:

A colloid is a solution that remains evenly distributed throughout the solution. These are also known as colloidal dispersions because the substances remain dispersed and do not settle to the bottom of the container.

- Colloidal state is characterized by the particle size of 1 to 100nm.
- When the particle size in <1nm, it is in true solution.
- For particle size >100nm, the matter exists as a visible precipitate.
- Thus, the colloidal state is an intermediate between true solution and precipitate.

Phases of Colloids:

Colloidal state is heterogeneous with two phases.

- Dispersed phase (internal phase) which constitutes the colloidal particles.
- Dispersion medium (external phase) which refer to the medium in which the colloidal particles are suspended.

Dispersion Medium	Dispersed Phase	Type of Colloid	Example
Solid	Solid	Solid sol	Ruby glass
Selid	Liquid	Solid emulsion/gel	Pearl, cheese
Solid	Ckas	Solid foam	Lava, pamice
Liquid .	Solid	Sol	Points, cell floids
Liquid	Liquid	Emulsion	Milk, oil in water
Liquid	Gas	Foam	Soap suds, whipped cream
Gas	Solid	Accord	Samolor
Gas	Liquid	Agresol	Fog. must

Classification of Colloids:

Based on the affinity of dispersion medium with dispersed phase, colloids are classified as lyophobic and lyophilic colloids.

- Lyophobic (solvent hating): These colloids do not have any affinity towards dispersion medium. When
 water is used as dispersion medium, the colloids are referred to as hydrophobic.
- Lyophilic (solvent loving): These colloids have distinct affinity towards dispersion medium. The term hydrophilic is used for the colloids when water is the dispersion medium.

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