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exothermic endothermic b how much heat will be absorbed or released if 10 g of NO is reacted with 10 g of H_2O

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phase what is the heat of fusion formula $h_n = h_{\text{fus}}$ what is the heat of vaporization formula $h_n = h_{\text{vap}}$ what is on the x axis of the
heating curve time

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option c is incorrect because 566 kJ/mol is the Δh per mole of O_2 and not for 2 moles even though options b and d are
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according to the following equation how much heat energy must be expelled by the body to rid the body of one pound of fat
 $2\text{C(s)} + 3\text{O}_2\text{(g)} \rightarrow 2\text{CO}_2\text{(g)}$ 7914 kJ is this endothermic or exothermic 2

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