

oscillator. Chapter 2 contains a discussion of average values and matrices, while the theory of perturbation, Dirac's theory of the radiation field, and related problems are discussed in Chapter 4. Chapters 5, 6, and 7 deal with the hydrogen atom, the structure of atoms containing many electrons, and multiplet structure. The Heitler-London method and molecular orbital method of treatment of the hydrogen molecule is the topic of Chapter 8, and in the following chapter, entitled *Interatomic and Intermolecular Forces*, ionic and covalent bindings are discussed.

All the new quantum mechanics concepts thus developed are applied in the following chapters to a consideration of the properties of matter: those of the metallic state (Chapter 10), the mechanical, chemical, and thermal properties (Chapter 11), electrical conductivity (Chapter 12), the nature of dielectrics (Chapter 13), and magnetism (Chapter 14).

In each case, the treatment begins in an elementary manner and is carried through to quite an advanced stage without the use of difficult mathematics. However, many of the topics which must necessarily be treated at a higher mathematical level for the sake of more detailed explanation are dealt with in 22 appendices (pages 435-516).

At the end of each chapter is a large selection of problems which should prove extremely useful to the student.

No review can really do justice to the clear and scholarly discussion in this treatise, which should be a "must" for every student who desires to become familiar with the many recent developments in the applications of the quantum theory to an understanding of the properties and behavior of matter in the solid state.

GENERAL ELECTRIC COMPANY
Schenectady, New York

SAUL DUSHMAN

● AN ADVANCED TREATISE ON PHYSICAL CHEMISTRY. VOLUME II

J. E. Partington, Professor of Chemistry in the University of London. Longmans, Green and Co., New York, 1951. xlv + 448 pp. 2 tables. Illustrated. 16 × 25.5 cm. \$10.

This second volume of Professor Partington's treatise contains the most comprehensive survey yet made of the properties of liquids and the experimental methods used in investigating the liquid state of matter. The subjects are discussed in the following order: *Densities* (pages 1-35), *Thermal Expansion* (pages 35-57), *Compressibilities* (pages 57-70), *Viscosities* (pages 70-127), *Thermal Conductivity* (pages 127-34), *Surface Tension* (pages 134-207), *Specific Heats* (pages 207-21), *Vapour Pressures* (pages 226-74), *Boiling Points* (pages 275-303), *Latent Heat of Vaporization* (pages 303-76).

Since there exists no exact theory which encompasses the wide variations in the behavior of the properties of liquids, the comprehensive critical survey contained in this volume is of great value. The thoroughness with which the experimental methods are described, the data recorded, and the thousands of references listed is an outstanding achievement. As illustrations of this thoroughness, one finds over 50 different empirical formulas representing the variation of vapor pressure with temperature and 60 or more empirical equations for the variation of the latent heat of vaporization with temperature.

Pages 377-428 contain a mathematical appendix in which the elements of the subjects, determinants, matrices, group theory, and co-ordinate geometry of three dimensions are treated in a lucid manner.

This reviewer has already gained much from reading carefully parts of this volume and would advise teachers and advanced students of general physical chemistry to arrange that this work be easily accessible.

YALE UNIVERSITY
New Haven, Connecticut

HERBERT S. HARNED

● THE STRUCTURE OF PHYSICAL CHEMISTRY

C. N. Hinshelwood, Dr. Lee's Professor of Chemistry in the University of Oxford. Oxford University Press, London, 1951. viii + 476 pp. 48 figs. 16.5 × 24.5 cm. \$7.

This book is an attempt to provide a synthesis of physical chemistry, such that diverse concepts, which are usually presented independently of one another, shall appear as parts of a systematically developed and unified structure. This structure is developed from the ideas of the atomic theory, considering first the simplest and most general concepts, and proceeding to increasingly complicated and detailed ones. The book is organized in six parts, each of which is developed in two to five chapters preceded by a synopsis of the part.

Part I, on the world as molecular chaos, presents the elementary ideas of the atomic theory and develops their consequences. The molecules are considered to be microscopic masses in chaotic motion, following the laws of classical mechanics. The interplay between the as yet unspecified forces of mutual attraction and repulsion determines the combinations that are possible and the states of aggregation in which the molecules exist. The concepts of entropy and of equilibria in chemical and physical changes flow then from these ideas.

Part II, on the control of the chaos by the quantum laws reshapes and refines the ideas of Part I to conform to the requirements of the quantum mechanics that is found to apply to molecular systems. Such systems are no longer considered to be small-scale replicas of macroscopic objects, but to be subject to qualitatively distinct rules. The forces among the parts of the system are still unspecified and the emphasis is still on the effects on the chaos of the general laws that control molecules.

Part III, on the electrical basis of matter, is concerned with the more detailed application of quantum mechanics to the structures of atoms and molecules.

Part IV, on forces, follows as a natural continuation of the preceding part. It is a discussion of the forces within nuclei, atoms, molecules, and macroscopic systems. Three are the forces that were left unspecified in the earlier parts of the book.

Part V, on the forms of matter in equilibrium, is concerned with the application of the concepts developed in the first four parts of the book to those systems that are of special interest to chemists. Metals, ionic and covalent compounds, crystal lattices, liquids and allied states of matter, and highly dispersed systems are all discussed from the common viewpoint—the effect of the ordering forces upon a chaotic system.

Part VI, on the passage towards equilibrium, is in many respects the climax of the book and is the longest single part (123 pages). It is on the rates of changes and the factors that influence these rates, and is developed from the quantum statistics that was discussed earlier. The book ends with an interesting and suggestive, if necessarily incomplete and speculative, application of the concepts to biological systems.

The treatment is on an advanced level as to the ideas, but only moderately so as to the mathematics and the language. The reasoning and the expression are, in fact, so clear and so attractive that the result is a pleasure to read. Derivations seem to be given principally to display the relationships of the ideas to one another and to clarify the ideas themselves. Consequently, one must know something of the subject in order to follow much of the book. The title correctly suggests that this book is neither a text nor a formal introduction to the subject. It is, rather, a humanistic attempt to understand and unify the subject by a treatment written for those who already have some knowledge of it.

Throughout the book there is no evident hesitation in including an idea even though it is difficult to grasp, nor in using any mathematical treatment that will clarify a concept. One of the merits of the treatment is that the author has avoided some usual, but misleading, analogies. "Some of the current working notions are expressed in words which easily become invested with a more literally descriptive character than they deserve, and many young chemists—this is my impression at least—are led to think they understand things which in fact

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JG Myers



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An Advanced Treatise on Physical Chemistry, 1955 Fluoroplastics, Volume 2: Melt Processible Fluoroplastics Sina Ebnesajjad, 2002-10-30 This is the second of a two volume series of books about fluoroplastics Volume 1 covers the non melt processible homopolymers requiring non traditional processing techniques Volume 2 is devoted to the melt processible fluoropolymers their polymerization and fabrication techniques including injection molding wire tube and film extrusion rotational molding blow molding compression molding and transfer molding Both a source of data and a reference the properties characteristics applications safety disposal and recycling of melt processible fluoropolymers are comprehensively detailed for immediate use by today s practicing engineering and scientists in the plastics industry Students will benefit from the book s arrangement and extensive references *JANAF Thermochemical Tables ...* Dow Chemical Company. Thermal Research Laboratory, Daniel Richard Stull, 1965 WADC Technical Report United States. Wright Air Development Division, 1956 Handbook of Solvents, Volume 2 George Wypych, 2024-02-11 This 4th edition of Handbook of Solvents Volume 2 contains the most comprehensive information ever published on solvents as well as an extensive analysis of the principles of solvent selection and use The book begins with a discussion of solvents used in over 30 industries which are the main consumers of solvents The analysis is conducted based on the available data and contains information on the types and frequently amounts of solvents used and potential problems and solutions Picking up where Handbook of Solvents Volume 1 leaves off Handbook of Solvents Volume 2 provides information on the methods of analysis of solvents and materials containing solvents with 2 sections containing standard and special methods of solvent analysis followed by a discussion of residual solvents left in the final products The environmental impact of solvents such as their fate and movement in the water soil and air fate based management of solvent containing wastes and ecotoxicological effects are discussed as are solvents impact on tropospheric air pollution The next 2 chapters are devoted to the toxicology of solvents and regulations aiming to keep solvent toxicity under control The analysis of the concentration of solvents in more than 15 industries specific issues related to the paint industry and characteristics of the environment in automotive collision repair shops are followed by a thorough discussion of regulations in the USA and Europe Following chapters show examples of solvent substitution by safer materials with an emphasis on supercritical solvents ionic liquids deep eutectic solvents and agriculture based products such as ethyl lactate Discussion of solvent recycling removal and degradation includes absorptive solvent recovery comparison of results of recovery and incineration and application of solar photocatalytic oxidation The book concludes with an evaluation of methods of natural attenuation of various solvents in soils and modern methods of cleaning contaminated soils Assists in solvent selection by providing key information and insight on environmental and safety issues Provides essential best practice guidance for human health consideration Discusses the latest advances and trends in solvent technology including modern methods of cleaning contaminated soils selection of gloves suits and respirators Electrochemistry and Corrosion Science

Nestor Perez,2007-05-08 Electrochemistry and Corrosion Science is a graduate level text professional reference that describes the types of corrosion on metallic materials The focus will be on modeling and engineering approximation schemes that describe the thermodynamics and kinetics of electrochemical systems The principles of corrosion behavior and metal recovery are succinctly described with the aid of pictures figures graphs and schematic models followed by derivation of equations to quantify relevant parameters Example problems are included to illustrate the application of electrochemical concepts and mathematics for solving complex corrosion problems This book differs from others in that the subject matter is organized around the modeling and predicating approaches that are used to determine detrimental and beneficial electrochemical events Thus this book will take a more practical approach and make it especially useful as a basic text and reference for professional engineers

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