### បំ halodoc

# 5 Cara Menerapkan Diet Intermittent Fasting bagi Pemula

- Puasa 12 jam sehari.

  Lalu, makan secara

  normal seperti puasa

  pada umumnya.
- Puasa 16 jam sehari.
  Disebut dengan metode
  16:8 dengan waktu makan
  pada 8 jam. Pria puasa
  selama 16 jam, sedangkan
  wanita 14 jam sehari.
- Puasa 2 hari dalam satu
  minggu. Disebut dengan
  metode 5:2, mengonsumsi
  makan sehat dengan porsi
  standar selama lima hari, lalu
  mengurangi asupan kalori
  selama dua hari.
- Puasa alternatif. Metode
  puasa setiap hari tanpa
  makanan padat atau
  maksimal 500 kalori untuk
  satu hari.

Puasa 24 jam dalam satu minggu. Berpuasa penuh selama satu atau dua hari dalam satu minggu atau nama lainnya eat-to-eat.

## **Tutorial Intermittent Fasting**

**David Scott Kingston** 

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web in today s experiment you will determine the solubility product ksp of calcium hydroxide ca oh 2 by measuring the concentration of ca oh 2 in a saturated solution calcium hydroxide is a sparingly soluble salt that dissolves according to the following reaction the solubility product expression for this reaction is

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web use the concentration of calcium hydroxide to calculate the concentration of calcium ions and hydroxide ions in the solution ca oh 2 à ca 2 2oh 5 x 10 2 m calcium ions 5 x 10 2 m 2 0 m hydroxide ions calculate calcium hydroxide ksp using the concentrations of calcium ions and hydroxide ions

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oh 2 m ca in saturated ca oh 2 m ksp of ca oh 2 average ksp of ca oh 2

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web the chemical equation used was ca oh 2 s ca2 aq 2oh aq the solubility product was determined to be ksp ca2 oh from the titration which was made by dripping a concentrated solution of calcium hydroxide into hydrochloric acid the calcium hydroxide was filtered before starting the titration

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web jul 12 2023 because the concentration of a pure solid such as ca 3 po 4 2 is a constant it does not appear explicitly in the equilibrium constant expression the equilibrium constant expression for the dissolution of calcium phosphate is therefore k ca2 3 po 3 4 2 ca3 po 4 2 ca3 po 4 2 k ksp ca2 3 po 3 4 2

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web therefore it was determined through error calculations and ease of usethat the most reliable valid method to use to get the closest ksp value tothe theoretical ksp for ca was the determination through ph method 2 introduction calcium hydroxide is a strong base soluble in water

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