

BASIC CHEMICAL THERMODYNAMICS

Sixth Edition

E. Brian Smith

Imperial College Press

Basic Chemical Thermodynamics

Michael Seilmaier

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

Basic Chemical Thermodynamics:

Basic Chemical Thermodynamics (6th Edition) E Brian Smith, 2013-10-04 This widely acclaimed text now in its sixth edition and translated into many languages continues to present a clear simple and concise introduction to chemical thermodynamics An examination of equilibrium in the everyday world of mechanical objects provides a starting point for an accessible account of the factors that determine equilibrium in chemical systems This straightforward approach leads students to a thorough understanding of the basic principles of thermodynamics which are then applied to a wide range of physical chemical systems The book also discusses the problems of non ideal solutions and the concept of activity and provides an introduction to the molecular basis of thermodynamics Over six editions the views of teachers of the subject and their students have been incorporated Reference to the phase rule has been included in this edition and the notation has been revised to conform to current IUPAC recommendations Students taking courses in thermodynamics will continue to find this popular book an excellent introductory text

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Basic Chemical Thermodynamics E. Brian Smith, 1998 **Basic Chemical Thermodynamics** Eric Brian Smith, 2004 *Basic Chemical Thermodynamics (6th Edition)*. E. Brian Smith, 2013

Chemical Thermodynamics Irving M. Klotz, 1960 **Chemical Thermodynamics: Basic Theory and Methods** Irving Myron KLOTZ, 1958 **Basic Chemical Thermodynamics** Theodore Robert Bowie, 1966 Lectures on Basic Chemical Thermodynamics Henry Carrison Thomas, 1974 *Chemical Thermodynamics: Principles and Applications* J. Bevan Ott, Juliana Boerio-Goates, 2000-06-07 Chemical Thermodynamics Principles and Applications presents a thorough development of the principles of thermodynamics an old science to which the authors include the most modern applications along with those of importance in developing the science and those of historical interest The text is written in an informal but rigorous style including anecdotes about some of the great thermodynamicists with some of whom the authors have had a personal relationship and focuses on real systems in the discussion and figures in contrast to the generic examples that are

often used in other textbooks The book provides a basic review of thermodynamic principles equations and applications of broad interest It covers the development of thermodynamics as one of the pre eminent examples of an exact science A discussion of the standard state that emphasizes its significance and usefulness is also included as well as a more rigorous and indepth treatment of thermodynamics and discussions of a wider variety of applications than are found in more broadly based physical chemistry undergraduate textbooks Combined with its companion book Chemical Thermodynamics Advanced Applications the practicing scientist will have a complete reference set detailing chemical thermodynamics Outlines the development of the principles of thermodynamics including the most modern applications along with those of importance in developing the science and those of historical interest Provides a basic review of thermodynamic principles equations and applications of broad interest Treats thermodynamics as one of the preeminent examples of an exact science Provides a more rigorous and indepth treatment of thermodynamics and discussion of a wider variety of applications than are found in more broadly based physical chemistry undergraduate textbooks Includes examples in the text and exercises and problems at the end of each chapter to assist the student in learning the subject Provides a complete set of references to all sources of data and to supplementary reading sources

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