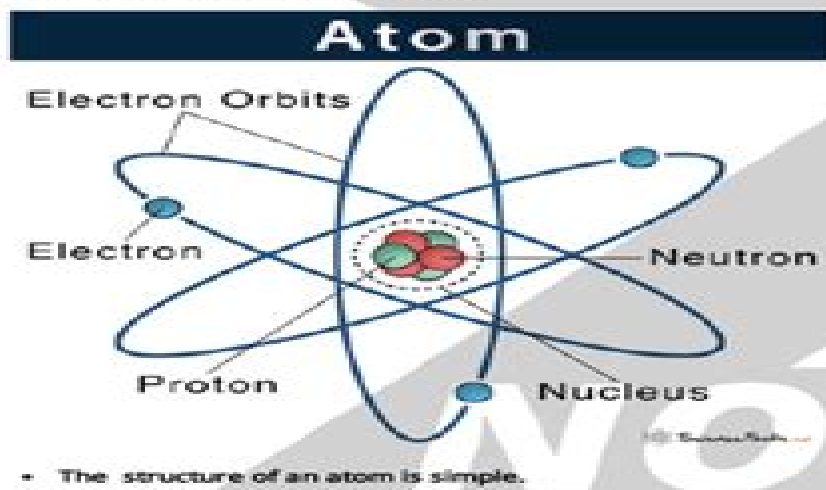


Component	Symbol	Function
Motor		A motor is a device that can convert electrical energy into mechanical energy.
Diode		A device which only allows current to flow in one direction.
Light Emitting Diode (LED)		A diode that emits light.

5. Atomic Physics

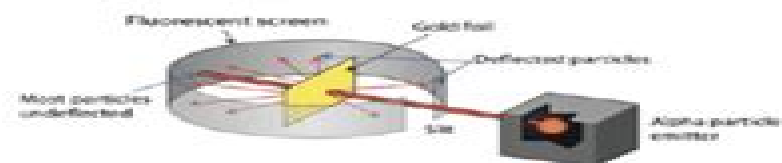
5.1. Nuclear Model of The Atom

- All matter is made up of atoms.



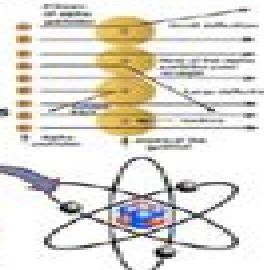
- The structure of an atom is simple.

Alpha Scattering Gold Foil experiment (Rutherford's)



Conclusions from a scattering experiment

- The vast majority of α particles were not deflected at all: **the atom must be mostly empty space.**
- Some α particles deflected through large angles: there must be a **very small nucleus with a positive charge with a large electric field near to its surface.**
- Alpha particles repelled: alpha particles are positively charged so the **nucleus must be positively charged in order to create an electrostatic force of repulsion.**
- Atoms are neutral overall: **electrons must be on the outside of the atom separating one atom from the next.**



The Atom

- Nucleus:** central part of atom made of protons (positively charged) and neutrons. These two types of particles are called nucleons. They are bound together by the strong nuclear force.
- Electrons:** almost massless particles which orbit nucleus in shells.
- Proton number:** number of protons in an atom.
- Nucleon number:** the number of nucleons (protons + neutrons) in an atom.
- The following is the nuclide notation for atoms.



ISOTOPES

- Atoms of the same element that have different numbers of neutrons e.g. Carbon 12 and Carbon 14.
- They have identical chemical properties but can have different physical properties eg: radioactive

Nuclear Fission

Nuclear fission is a reaction in which the nucleus of an atom splits into two or more smaller nuclei, because of the addition of an electron. The fission process often produces gamma photons, and releases a very large amount of energy.

Atomic And Space Physics

N Colangelo



Atomic And Space Physics:

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