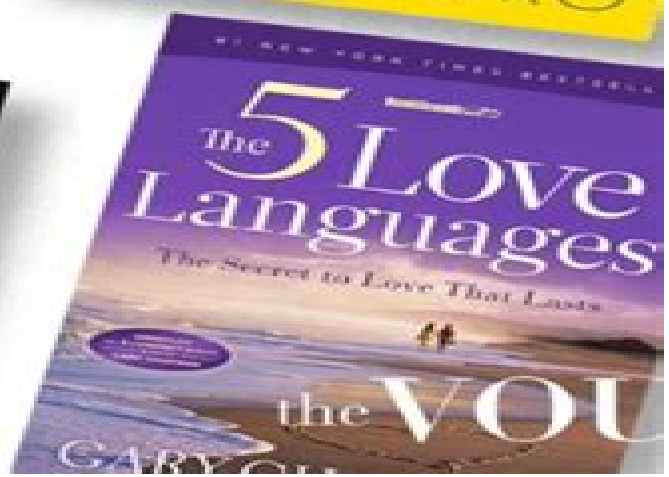
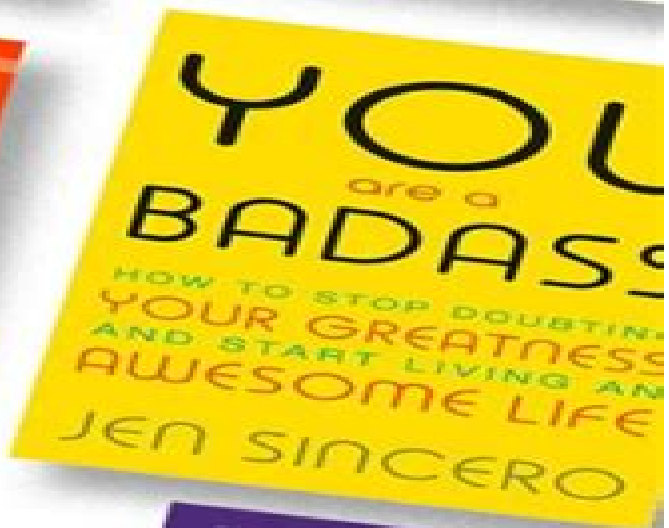
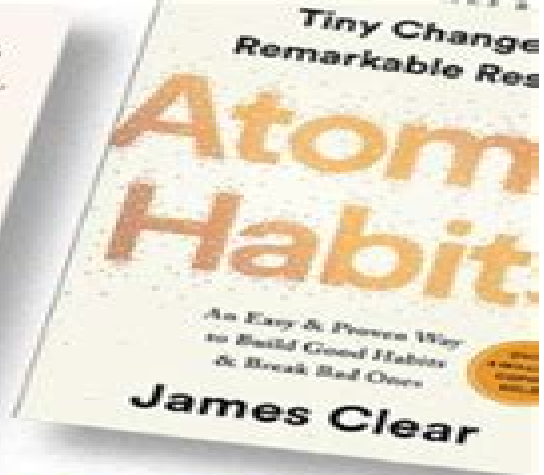
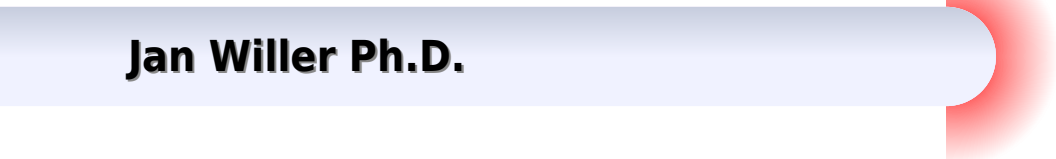




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web acid base titration difference between acid and base acids and bases definition an acid is any hydrogen containing substance that is capable of donating a proton hydrogen ion to another substance a base is a molecule or ion able to accept a hydrogen ion from an acid acidic substances are usually identified by their sour taste

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base than  $\text{Cl}^-$   $\text{pK}_a$  7.7 to identify the strongest base we can determine their weakest conjugate acid the conjugate acids of  $\text{CH}_3\text{NH}_2$  and  $\text{S}^{2-}$  are  $\text{CH}_4$ ,  $\text{NH}_3$  and  $\text{HS}^-$  respectively

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web acid base reactions are essential in both biochemistry and industrial chemistry moreover many of the substances we encounter in our homes the supermarket and the pharmacy are acids or bases for example aspirin is an acid acetylsalicylic acid and

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web nov 13 2022 the key to understanding acids as well as bases and salts had to await Michael Faraday's mid nineteenth century discovery that solutions of salts known as electrolytes conduct electricity this implies the existence of charged particles that can migrate under the influence of an electric field

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